

 LABORATORY PRODUCTS

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# SAFETY DATA SHEET Hi Valley Chemical

# **Ferric Oxide**

# PRODUCT AND COMPANY IDENTIFICATION

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# HAZARDS IDENTIFICATION

## **Classification of the Substance or Mixture**

#### GHS Classification in Accordance with 29 CFR 1910 (OSHA HCS):

Health, Skin corrosion/irritation, 2 Health, Serious Eye Damage/Eye Irritation, 2 A Health, Specific target organ toxicity - Single exposure, 2

## **GHS Label Elements, Including Precautionary Statements**

#### **GHS Signal Word: WARNING**

#### **GHS Hazard Pictograms:**

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#### **GHS Hazard Statements:**

- H315 Causes skin irritation
- H319 Causes serious eye irritation
- H371 May cause damage to organs

#### **GHS Precautionary Statements:**

- P261 Avoid breathing dust/fume/gas/mist/vapors/spray.
- P264 Wash face, hands and any exposed skin thoroughly after handling.
- P271 Use only outdoors or in a well-ventilated area.
- P280 Wear protective gloves/protective clothing/eye protection/face protection.

## Hazards not Otherwise Classified (HNOC) or not Covered by GHS

Inhalation:	May cause respiratory irritation
Skin Contact:	May cause irritation.
Eye Contact:	May cause irritation.

## COMPOSITION/INFORMATION ON INGREDIENTS

## Ingredients:

Cas#

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% Chemical Name

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## FIRST AID MEASURES

Inhalation:	If inhaled, move person to fresh air. If not breathing, give artificial respiration. Consult a physician.
Skin Contact:	Wash with soap and water.
Eye Contact:	Flush eyes with water as a precaution.
Ingestion:	Do NOT induce vomiting. Never give anything by mouth to an unconscious person. Rinse mouth with water. Consult a physician.

FIRE FIGHTING MEASURES

Extinguishing media Suitable extinguishing media Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.

Special hazards arising from the substance or mixture Iron oxides

Advice for firefighters Wear self-contained breathing apparatus for firefighting if necessary.

Further information No data

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# ACCIDENTAL RELEASE MEASURES

#### Personal precautions, protective equipment and emergency procedures:

Avoid dust formation. Wear respiratory protection. Avoid breathing vapors, mist or gas. Ensure adequate ventilation. Evacuate personnel

to safe areas.

#### Environmental precautions:

No special environmental precautions required.

## Methods and materials for containment and cleaning up:

Sweep up and shovel. Keep in suitable, closed containers for disposal.

7	HANDLING AND STORAGE
Handling Precautions:	Further processing of solid materials may result in the formation of combustible dusts. The potential for combustible dust formation should be taken into consideration before additional processing occurs.
Storage Requirements:	Provide appropriate exhaust ventilation at places where dust is formed. Keep container tightly closed. Store in cool/dry and well ventilated area.
8	EXPOSURE CONTROLS/PERSONAL PROTECTION
Engineering Controls:	Appropriate engineering controls: Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday.
Personal Protective Equipment:	Ferric oxide (1309-37-1) []
	Personal protective equipment
	Respiratory protection: For nuisance exposures use type P95 (US) or type P1 (EU EN 143) particle respirator.For higher level protection use type OV/AG/P99 (US) or type ABEK-P2 (EU EN 143) respirator cartridges. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU).

Hand protection: Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique (without touching gloves outer surface) to avoid skin contact with this product. Dispose of

contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands.

Eye protection: Safety glasses with side-shields conforming to EN166 Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU).

Skin and body protection: impervious clothing, The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace.

Hygiene measures: Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday.

Ferric oxide (1309-37-1) []

Components with workplace control parameters

TWA Pneumoco	5 mg/m3 oniosis Not c	USA. ACGIH Threshold Limit Values (TLV) lassifiable as a human carcinogen	
TWA	15 mg/m3	USA. Occupational Exposure Limits (OSHA) - Table Z- 1 Limits for Air Contaminants	
TWA	5 mg/m3	USA. Occupational Exposure Limits (OSHA) - Table Z- 1 Limits for Air Contaminants	
TWA	10 mg/m3	USA. Occupational Exposure Limits (OSHA) - Table Z- 1 Limits for Air Contaminants	
TWA	10 mg/m3	USA. OSHA - TABLE Z-1 Limits for Air Contaminants -	
See Appendix D - Substances with No Established RELs			
TWA	5 mg/m3	USA. NIOSH Recommended Exposure Limits	

PHY	SICAL ANI	CHEMICAL	PROPERTIES
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Appearance:	Red, orange, or reddish-brown powder		
Physical State:	Powder	Odor:	None
Odor Threshold:	No data available	Molecular Formula:	Fe2O3
Particle Size:	No data available	Solubility:	Insoluble in alcohol, cold water, diethyl et
Spec Grav./Density:	5.12-5.24	Softening Point:	No data available
Viscosity:	No data available.	Percent Volatile:	No data available
Sat. Vap. Conc.:	No data available	Heat Value:	No data available
Boiling Point:	No data available	Freezing/Melting Pt.:	1538 C (2800 F)
Flammability:	No data available.	Flash Point:	No data available
Partition Coefficient:	No data available	Octanol:	No data available
Vapor Pressure:	No data available	Vapor Density:	No data available
pH:	No data available	VOC:	No data available
Evap. Rate:	No data available.	Bulk Density:	No data available
Molecular weight:	159.69	Auto-Ignition Temp:	No data available
Decomp Temp:	No data available	UFL/LFL:	No data available

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## STABILITY AND REACTIVITY

Reactivity: Chemical Stability: Conditions to Avoid: Materials to Avoid: Hazardous Decomposition: Hazardous Polymerization: No data available Stable under recommended storage conditions. No data available Strong Acids; Peroxides; Chloroformates No data available No data available.

## **TOXICOLOGICAL INFORMATION**

Ferric oxide (1309-37-1) []

Information on toxicological effects

Acute toxicity: Oral LD50 no data available Inhalation LC50 Dermal LD50 Other information on acute toxicity

Skin corrosion/irritation: Skin - Human - Skin irritation

Serious eye damage/eye irritation: Eyes - Human - Eye irritation

Eyes - Human - Moderate eye irritation

Respiratory or skin sensitization: no data available

Germ cell mutagenicity: no data available

Carcinogenicity:

Carcinogenicity - rat - Subcutaneous:

Tumorigenic:Equivocal tumorigenic agent by RTECS criteria. Tumorigenic:Tumors at site or application. This product is or contains a component that is not classifiable as to its carcinogenicity based on its IARC, ACGIH, NTP, or EPA classification.

IARC: 3 - Group 3: Not classifiable as to its carcinogenicity to humans (Diiron trioxide) NTP: No component of this product present at levels greater than or equal to 0.1% is identified as a known or anticipated carcinogen by

NTP.

OSHA: No component of this product present at levels greater than or equal to 0.1% is identified as a carcinogen or potential carcinogen by OSHA.

Reproductive toxicity: no data available

Teratogenicity: no data available

Specific target organ toxicity - single exposure (Globally Harmonized System): Inhalation - May cause respiratory irritation.

Specific target organ toxicity - repeated exposure (Globally Harmonized System): no data available

Aspiration hazard: no data available

Potential health effects: Inhalation May be harmful if inhaled. Causes respiratory tract irritation. Ingestion May be harmful if swallowed. Skin May be harmful if absorbed through skin. Causes skin irritation. Eyes Causes eye irritation.

Signs and Symptoms of Exposure: Long term inhalation exposure to iron (oxide fume or dust) can cause siderosis. Siderosis is considered to be a benign pneumoconiosis and does not normally cause significant physiologic impairment. Siderosis can be observed on x-rays with the lungs having a mottled appearance., To the best of our knowledge, the chemical, physical, and toxicological properties have not been thoroughly investigated.

Synergistic effects: no data available

Additional Information:

RTECS: NO7400000

# ECOLOGICAL INFORMATION

Ferric oxide (1309-37-1) []

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Information on ecological effects

Toxicity: no data available

Persistence and degradability: no data available

Bioaccumulative potential: no data available

Mobility in soil: no data available

PBT and vPvB assessment: no data available

Other adverse effects: no data available

# DISPOSAL CONSIDERATIONS

Ferric oxide (1309-37-1) []

Waste treatment methods

Product: Offer surplus and non-recyclable solutions to a licensed disposal company. Contact a licensed professional waste disposal service to dispose of this material.

Contaminated packaging: Dispose of as unused product.

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## TRANSPORT INFORMATION

Non-hazardous for air, sea and road freight.

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## **REGULATORY INFORMATION**

Component (CAS#) [%] - CODES

Ferric oxide (1309-37-1) [n/a%] MASS, OSHAWAC, PA, TSCA, TXAIR

Regulatory CODE Descriptions

MASS = MA Massachusetts Hazardous Substances List OSHAWAC = OSHA Workplace Air Contaminants PA = PA Right-To-Know List of Hazardous Substances TSCA = Toxic Substances Control Act TXAIR = TX Air Contaminants with Health Effects Screening Level

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NFPA: Health = 2, Fire = 0, Reactivity = 0, Specific Hazard = n/aHMIS III: Health = 2, Fire = 0, Physical Hazard = 0



## Disclaimer:

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